

Accumulation of Contaminants in the Distribution Systems

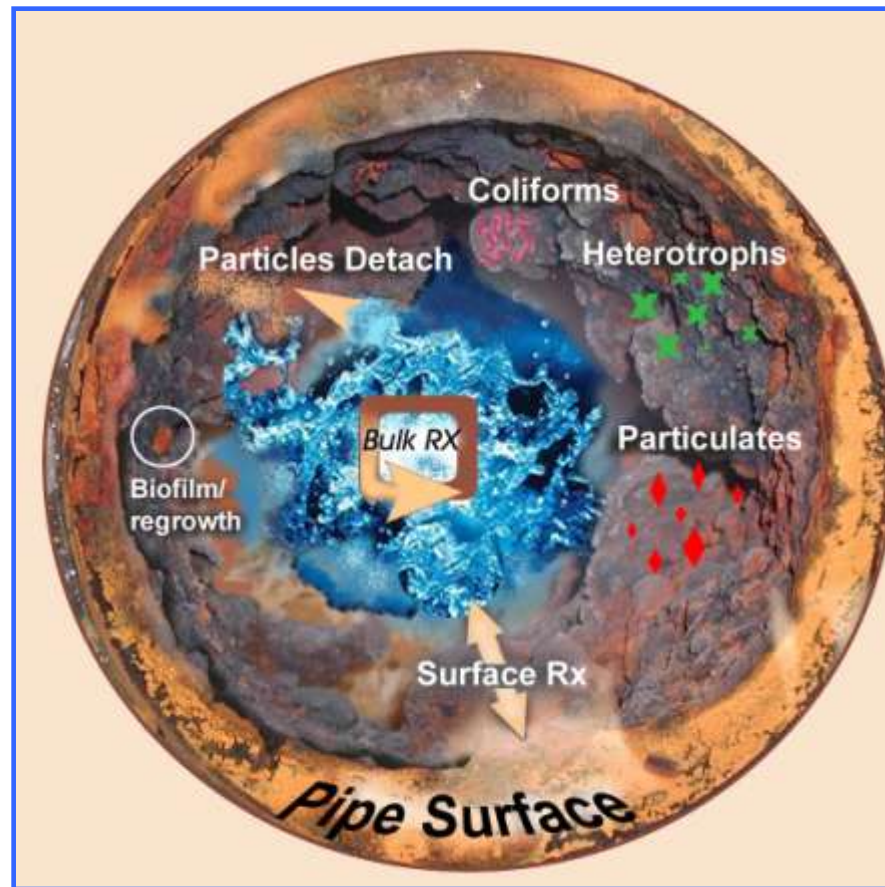
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Corroding Pipes are Complicated Reactors that can Accumulate Trace Contaminants



General Nature of Metallic Pipe Surfaces

- Oxides, hydroxides, hydroxycarbonates, carbonates, hydroxysulfates, etc. from corrosion
- Similar compounds from deposition or post-precipitation (particularly Fe, Mn, Al), may include silicates
- Phosphates from corrosion control
- All may be mixed with NOM
- Biofilms present in some areas and some materials



Practical Issues of Contaminant Accumulation

- What contaminants are involved (health risk?)
- What is the “equilibrium” mass of deposit?
- Where are the contaminants located
 - *Relative to consumer ingestion?*
 - *Relative to regulatory monitoring locations?*
 - *Relative to types of mains/pipes?*



Accumulation of Contaminants in the DS



Corrosion deposits, sediment, and other solids that collect in the DS can accumulate contaminants if in the water



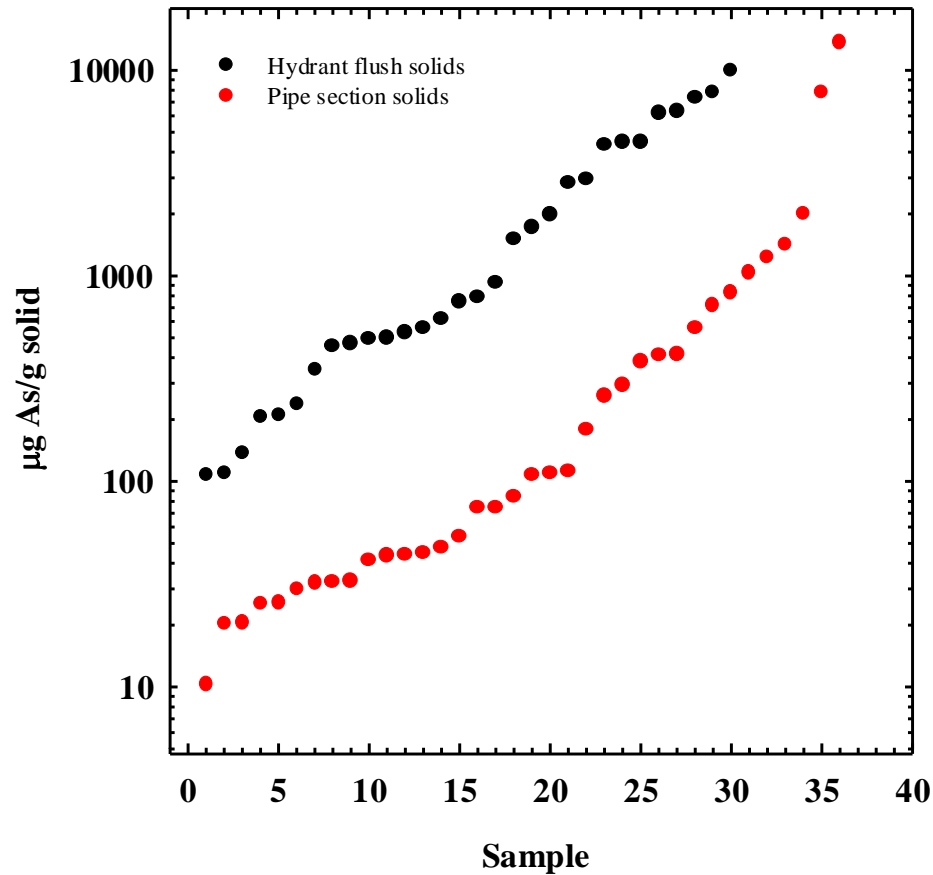
Solids Analysis

- Acid digestion/ICP-MS (Battelle)
 - *Ca, Mn, Fe, Mg, P, Si, As, others*
 - *Units*
- XRD
 - *Mineral phases*
- Electron microprobe-WDS (Battelle)
 - *Quantitative elemental mapping*
- SEM-Wavelength dispersive spectrometer- imaging and elemental mapping



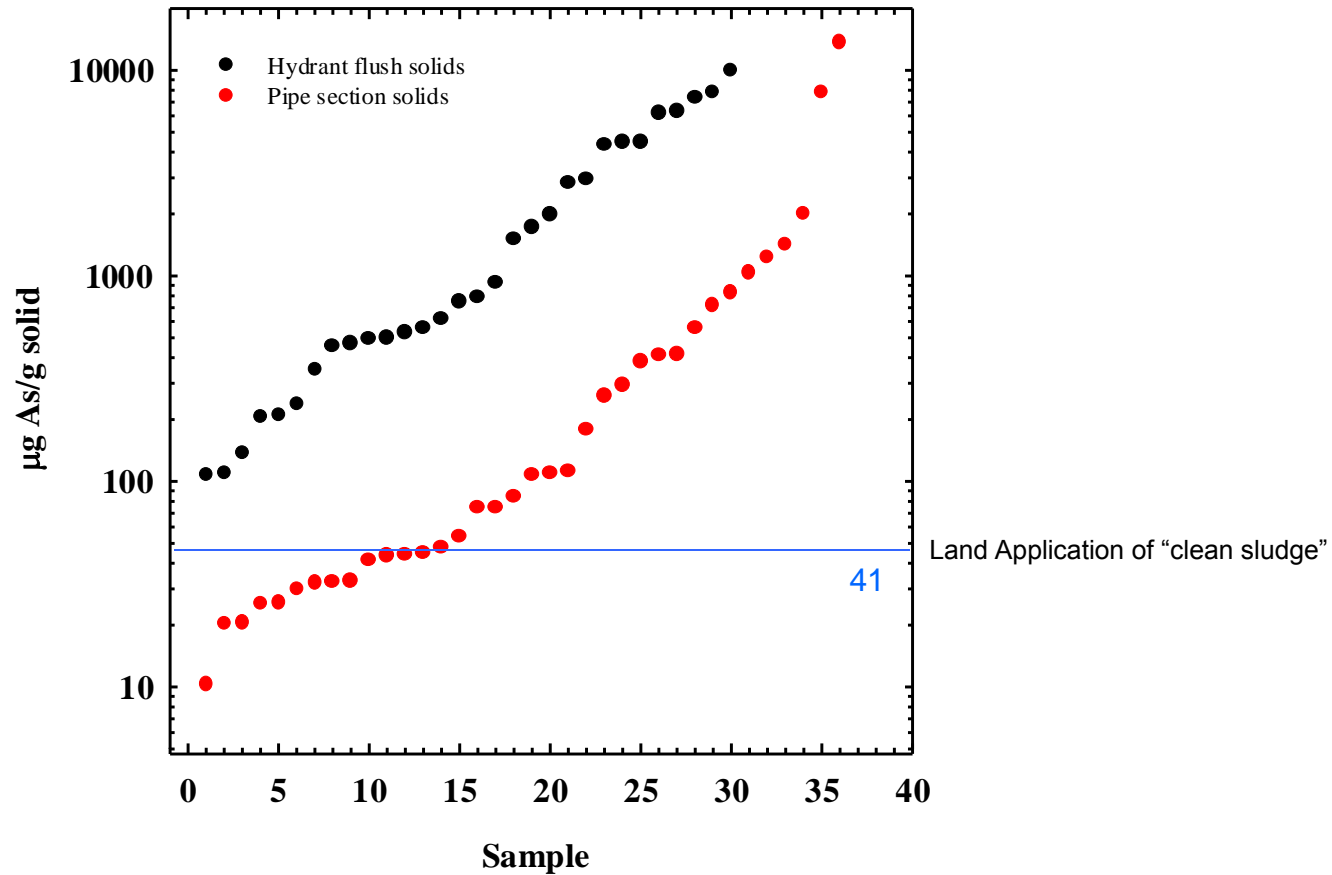
Arsenic Accumulation in the DS

ICP-MS Analysis



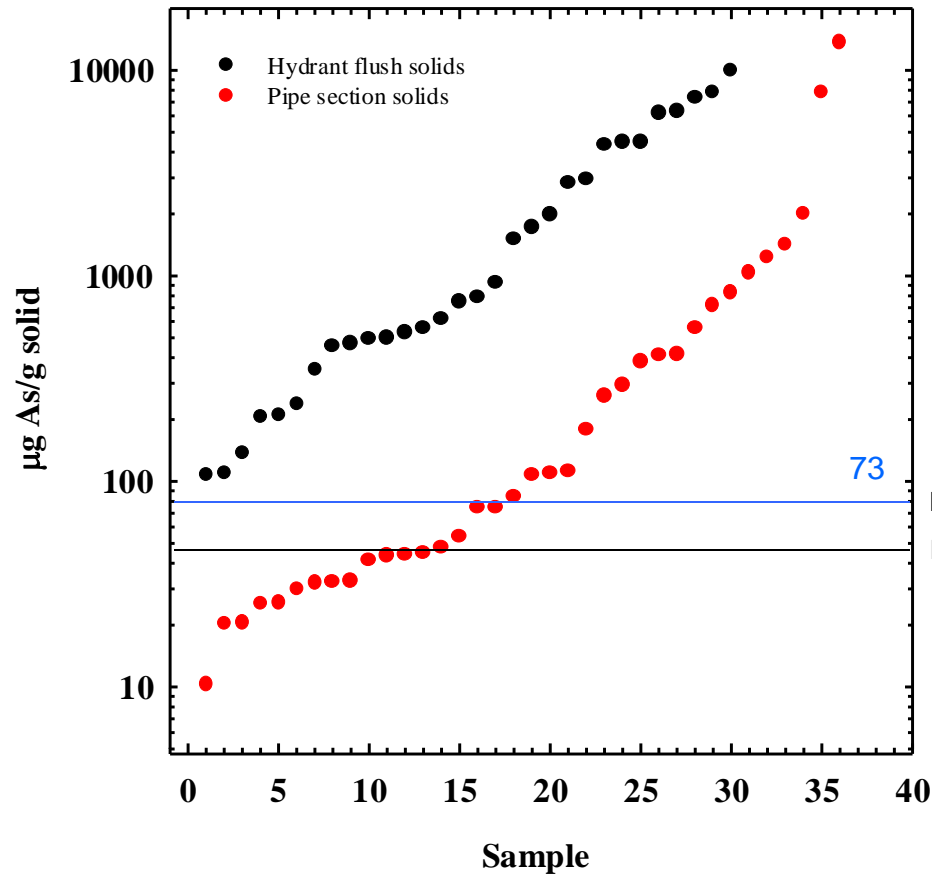
Arsenic Accumulation in the DS

ICP-MS Analysis



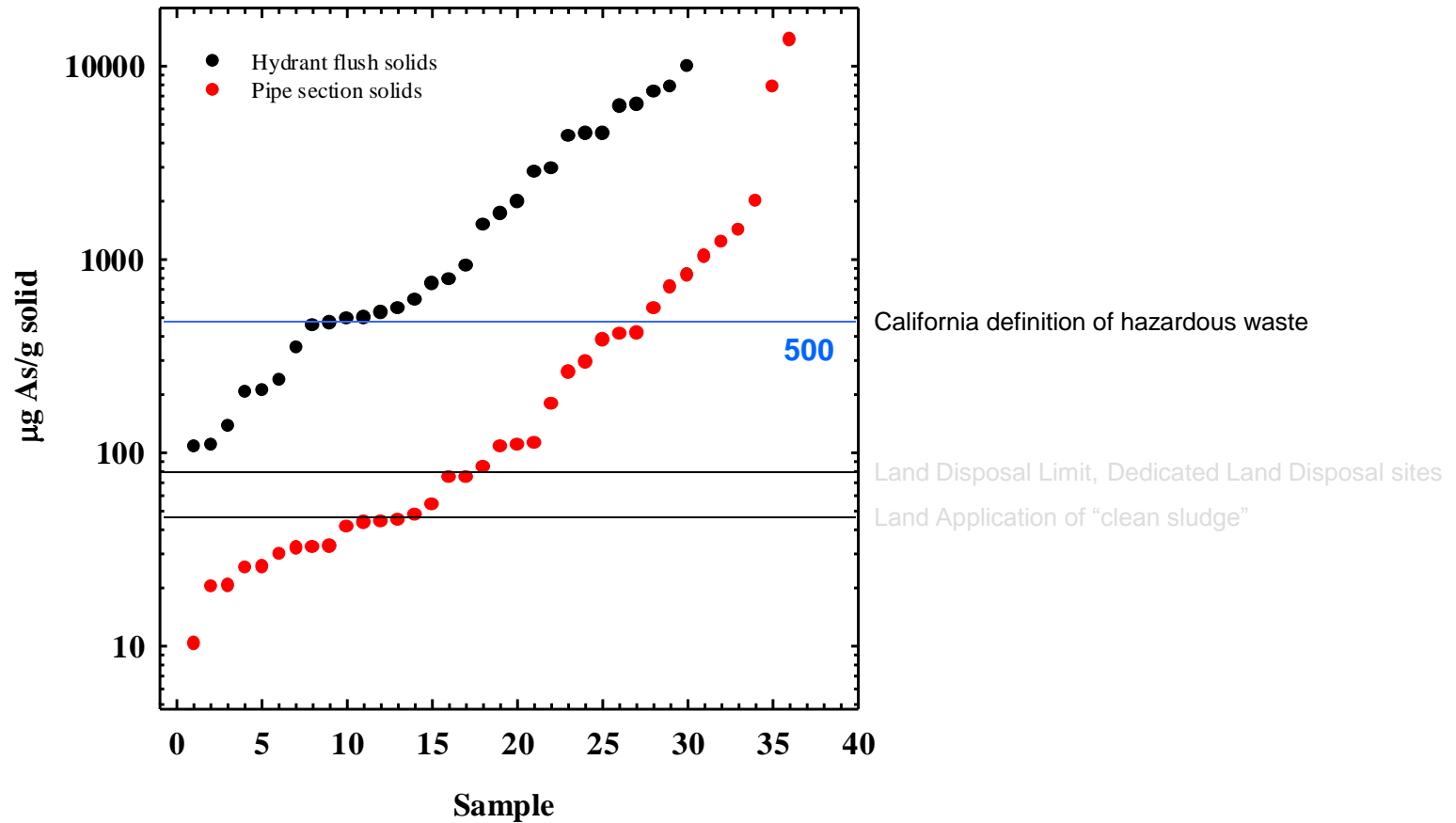
Arsenic Accumulation in the DS

ICP-MS Analysis



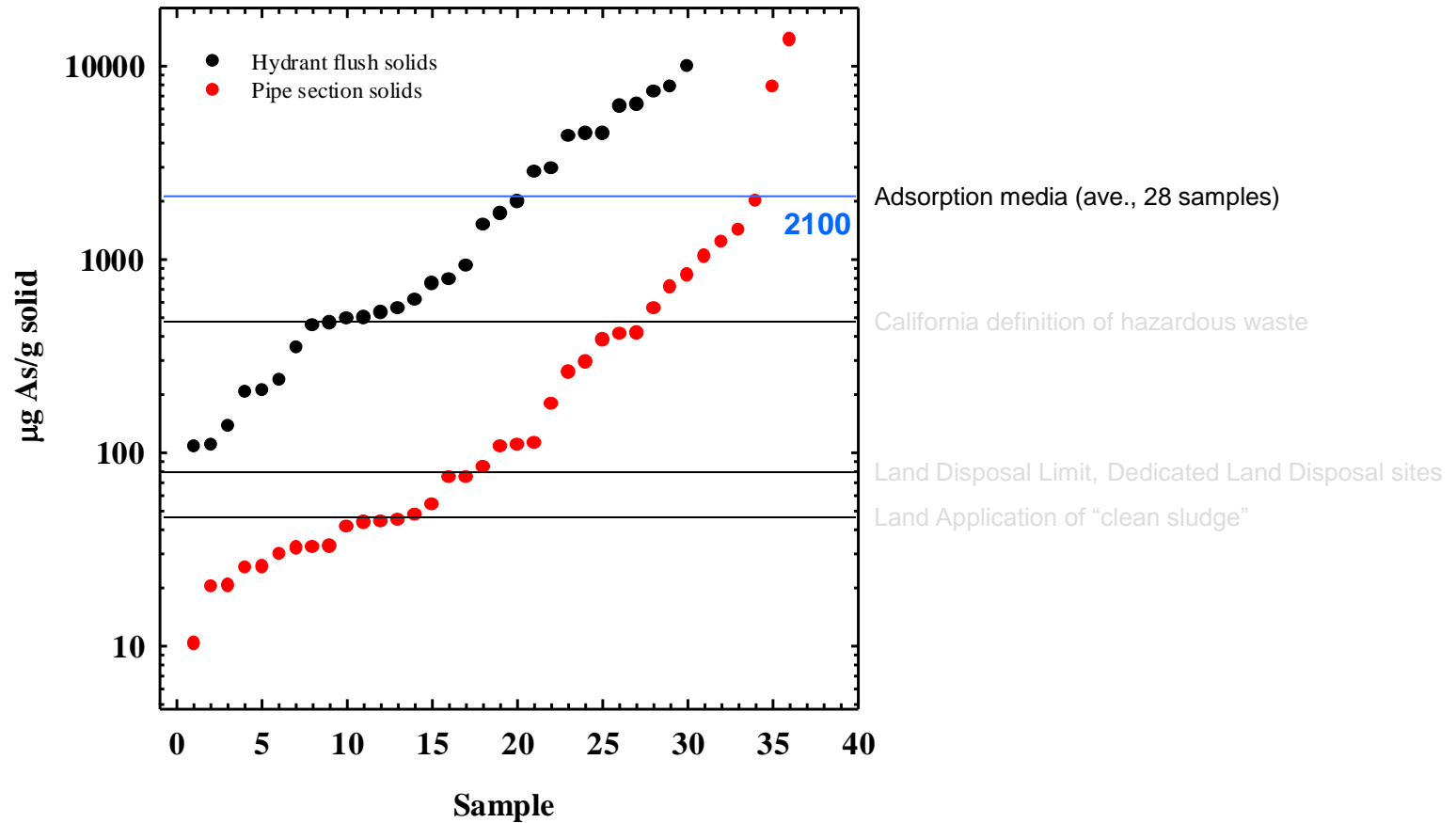
Arsenic Accumulation in the DS

ICP-MS Analysis



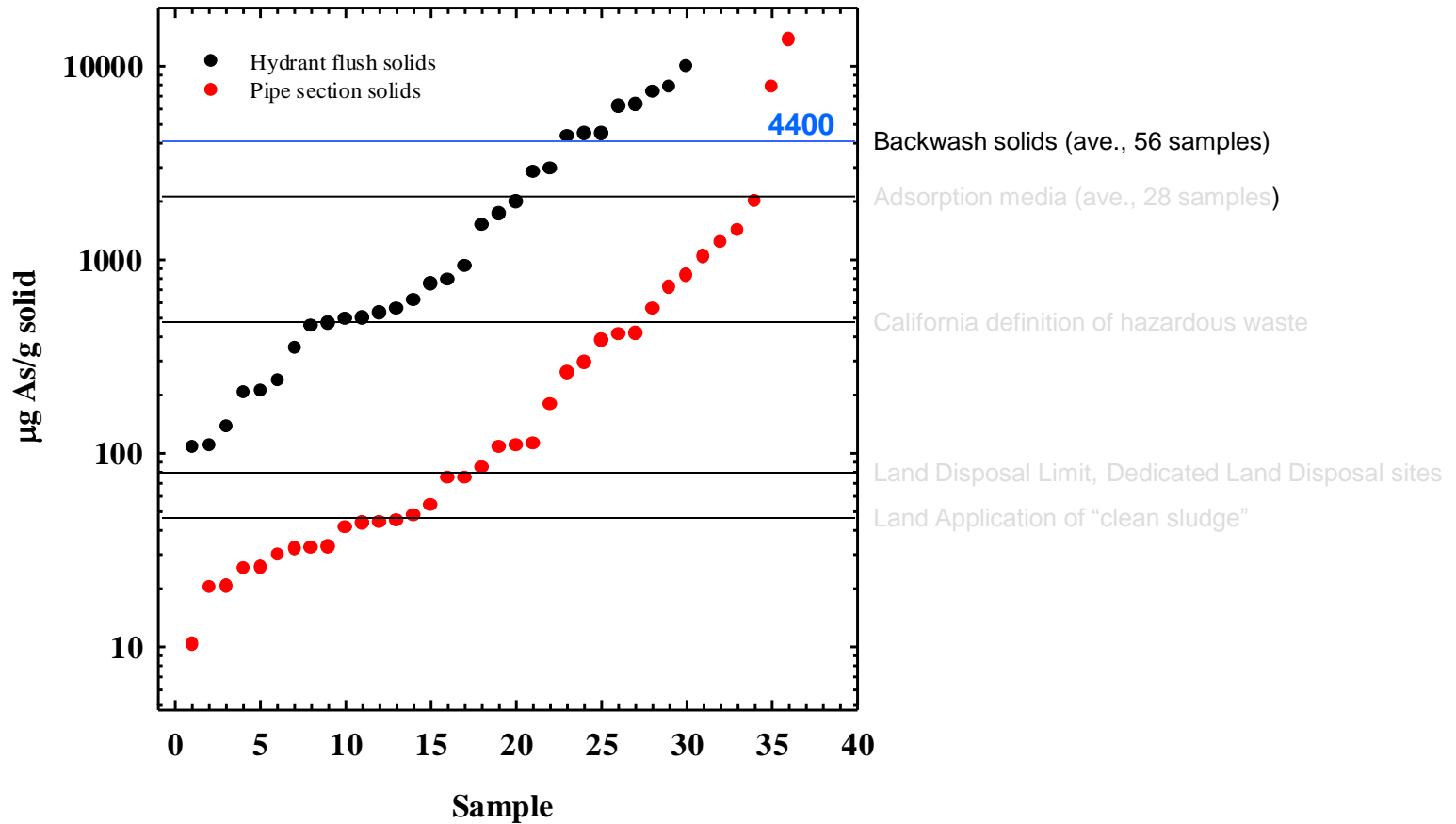
Arsenic Accumulation in the DS

ICP-MS Analysis

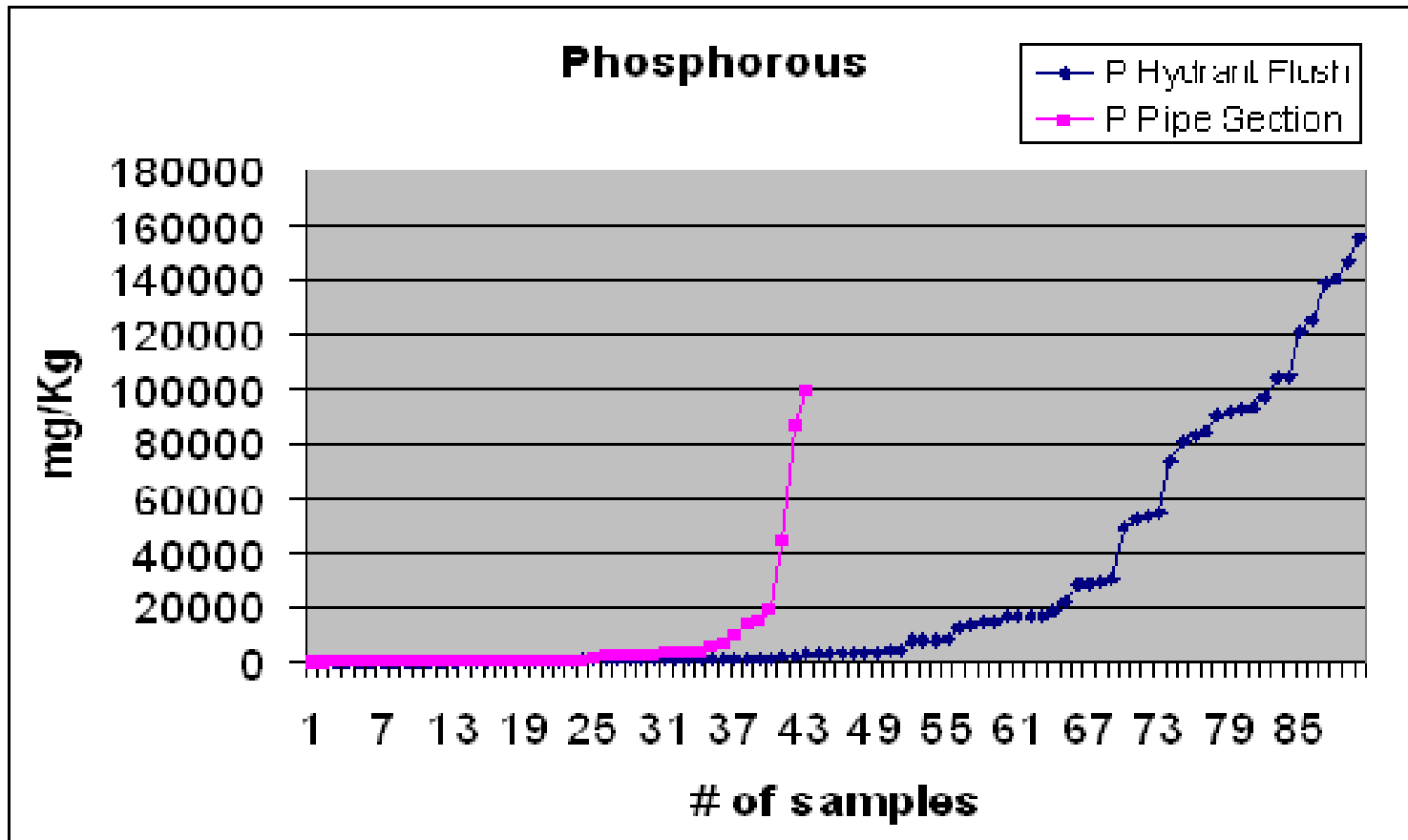


Arsenic Accumulation in the DS

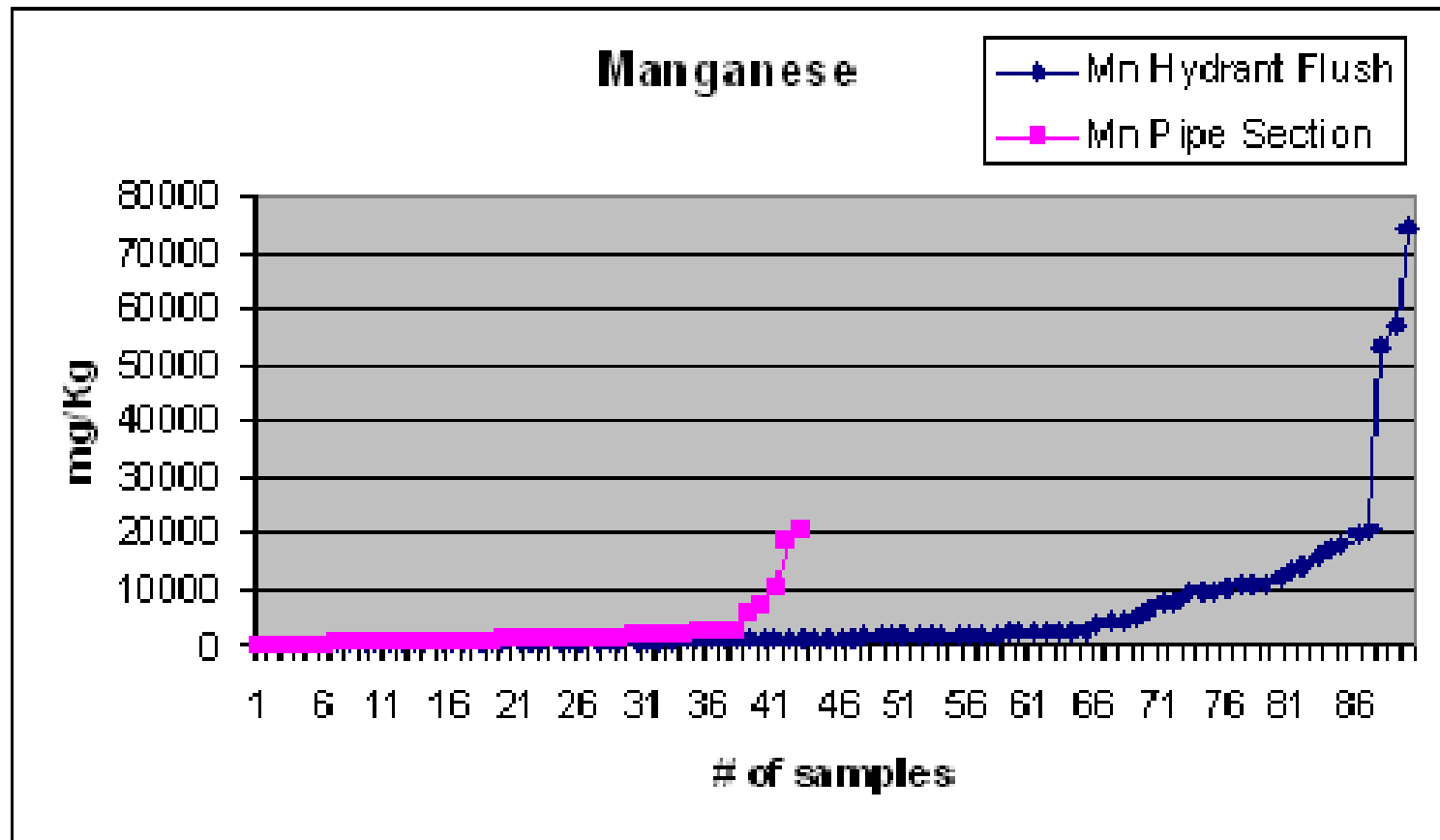
ICP-MS Analysis



Phosphorus Accumulation



Manganese Accumulation



Practical Issues of Contaminant Accumulation

- How easily is it destabilized?
 - *How is the contaminant bound?*
 - Solid mineral phase?
 - Sorbed?
 - *Phosphates: Do they*
 - “Seal” the surfaces?
 - Dissolve or displace the surface compounds and layers?
 - Mobilize sediment particles?
 - Any or all of the above?
 - *Hydraulic factors: pressure or flow changes*



Particle Mobilization

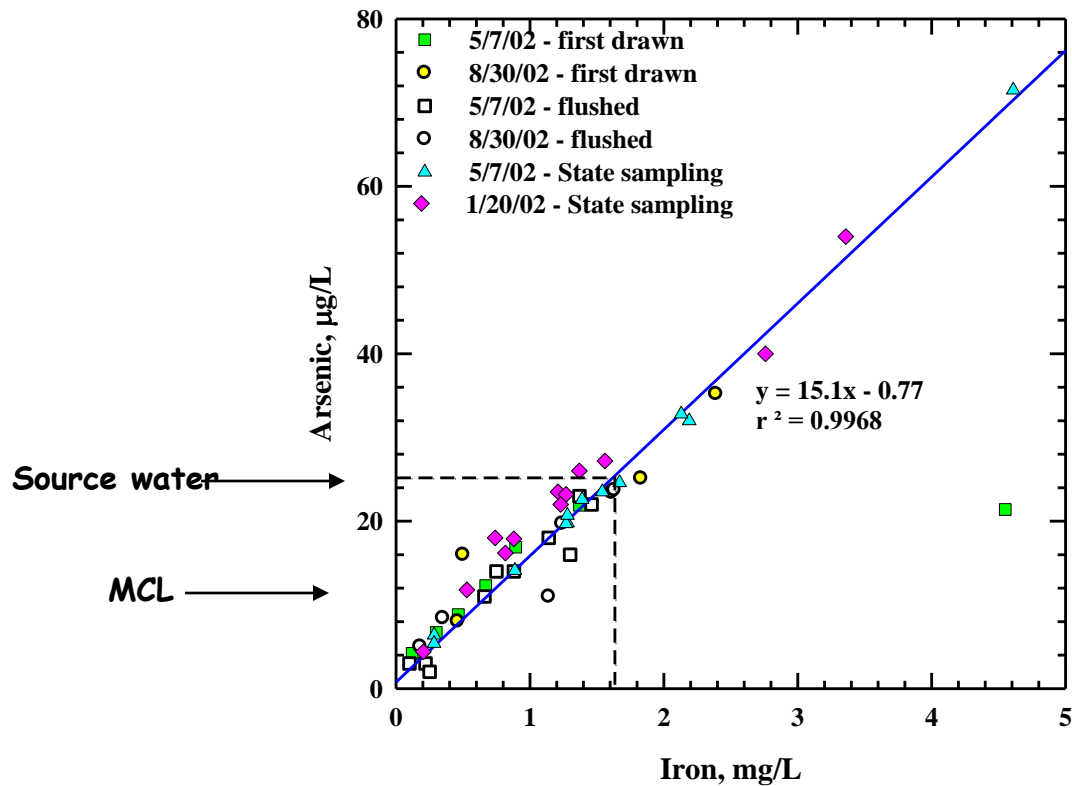
Case Study 1: Particulate Release of Arsenic in Distribution Systems

- Colored water events led to sampling and the finding that As levels ($>100 \mu\text{g/L}$)
- Also high iron levels ($>15 \text{ mg/L}$)
- Lawsuit and media attention
- 73 mg Ca/L , 32 mg Mg/L ,
 $17 \text{ mg SiO}_2/\text{L}$,
pH mid 7's
- $24 \mu\text{g As/L}$, 1.6 mg Fe/L
- Chlorination



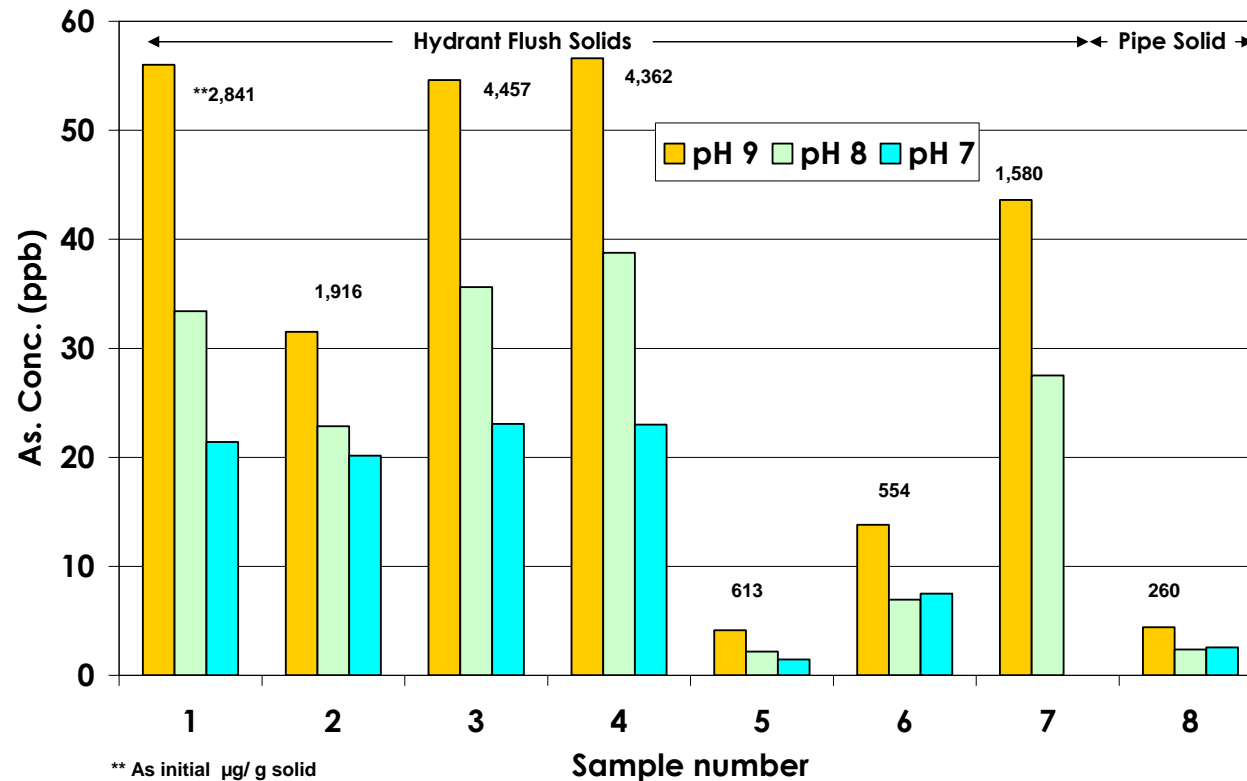
Particle Mobilization

Case Study 1: Relationship Between Arsenic and Iron in Distribution System Samples



Desorption

Research: Desorption from Drinking Water Distribution System Solids

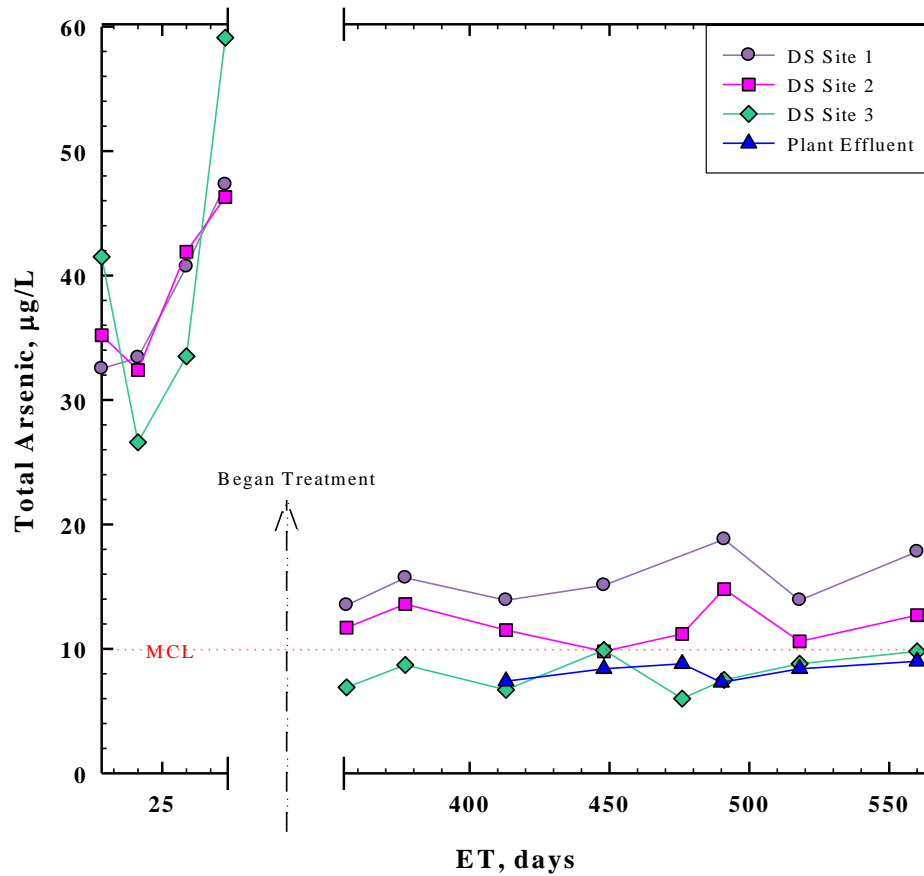


- Samples 1, 2,3,4 and 7 correspond to the same Utility
- The majority of these solids are hydrant flush material



Desorption

Lidgerwood, ND



Initial soluble As concentration:
125 µg/L (reduced to approximately 35 µg/L prior to DS)

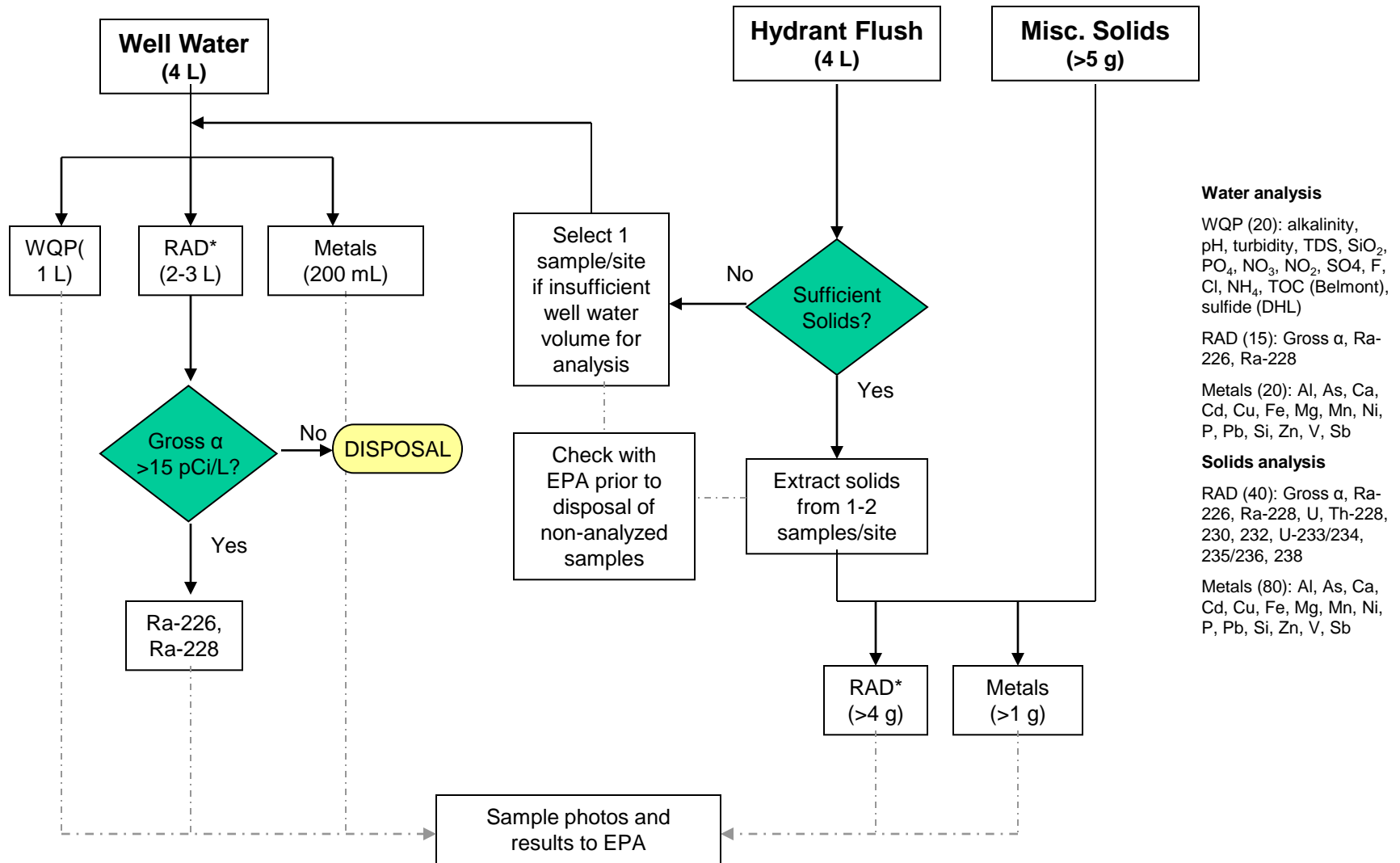


Accumulation of Radium and Uranium in Distribution Systems

- Battelle Contractor
- Pipe sections and fire hydrant flush samples
- Midwest, Texas water systems
- Status: samples concentrated currently being analyzed



Radium and Uranium Sample Flow Chart



Water analysis
 WQP (20): alkalinity, pH, turbidity, TDS, SiO₂, PO₄, NO₃, NO₂, SO₄, F, Cl, NH₄, TOC (Belmont), sulfide (DHL)

RAD (15): Gross α, Ra-226, Ra-228

Metals (20): Al, As, Ca, Cd, Cu, Fe, Mg, Mn, Ni, P, Pb, Si, Zn, V, Sb

Solids analysis
 RAD (40): Gross α, Ra-226, Ra-228, U, Th-228, 230, 232, U-233/234, 235/236, 238

Metals (80): Al, As, Ca, Cd, Cu, Fe, Mg, Mn, Ni, P, Pb, Si, Zn, V, Sb

WQP = water quality parameter analyses; RAD = radiological isotopes analyses; Metals = ICP/MS analyses; TBD = to be determined.

* = first priority for analysis.

All samples will be screened by Radiation Safety Services for proper handling and storage guidelines.



RESEARCH & DEVELOPMENT

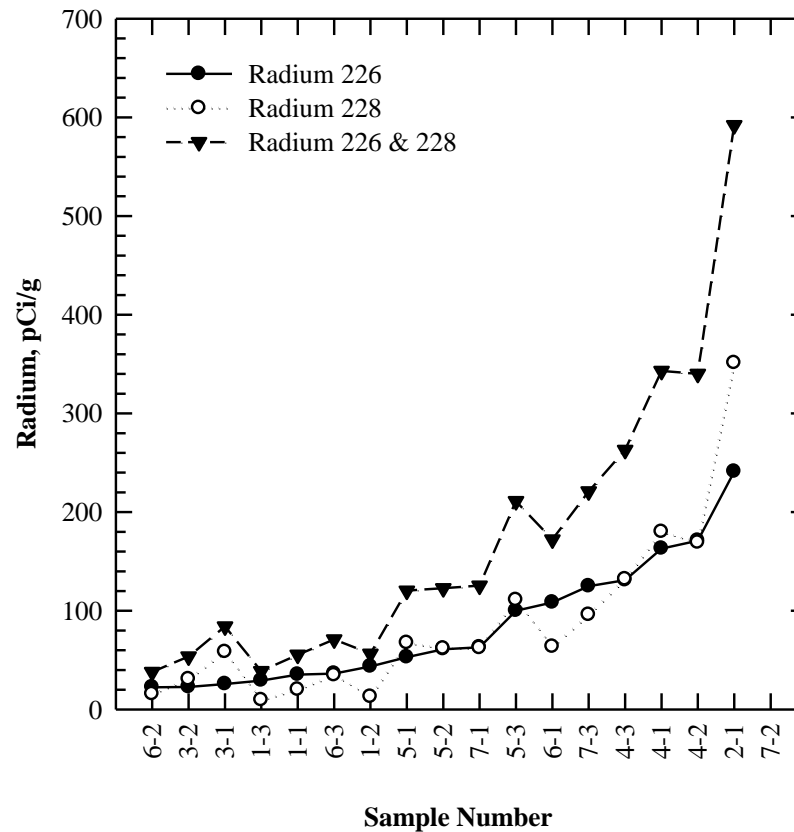
Building a scientific foundation for sound environmental decisions

Accumulation of Contaminants in the DS

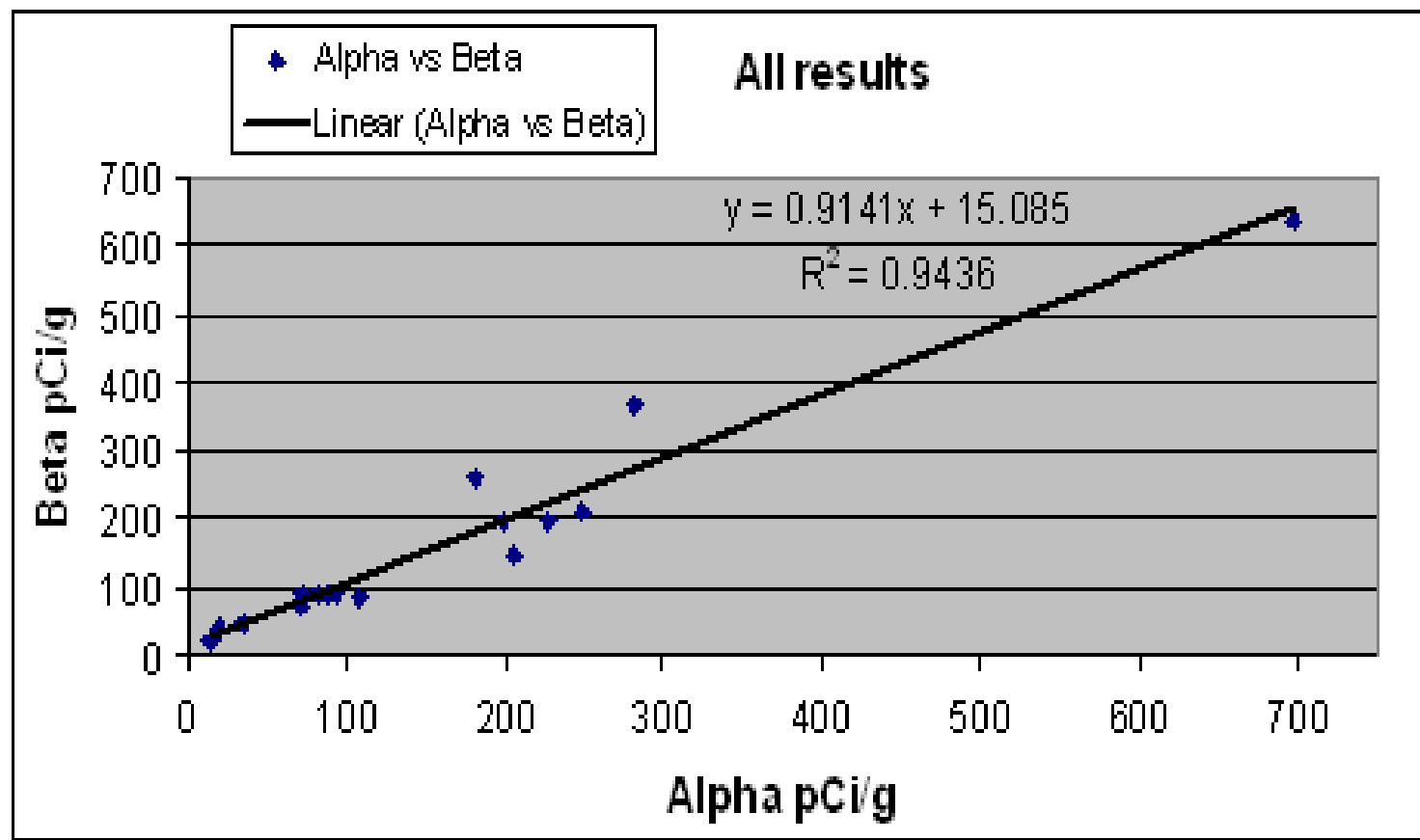
- Mg, Al, Si, P, Ca, V, Mn, Fe, Ni, Cu, Zn, As, Cd, Sb, Ba, Pb
- Gross α ; gross β ; Ra-226 & 228; Th- 228, 230 & 232; U- 233/234, 235, 236 & 238
- 18 sample from 7 location in Midwest
 - 13 to 698 pCi/g gross α
 - 22 to 637 pCi/g gross β
 - 23 to 241 pCi/g radium-226
 - 16 to 351 pCi/g radium-228



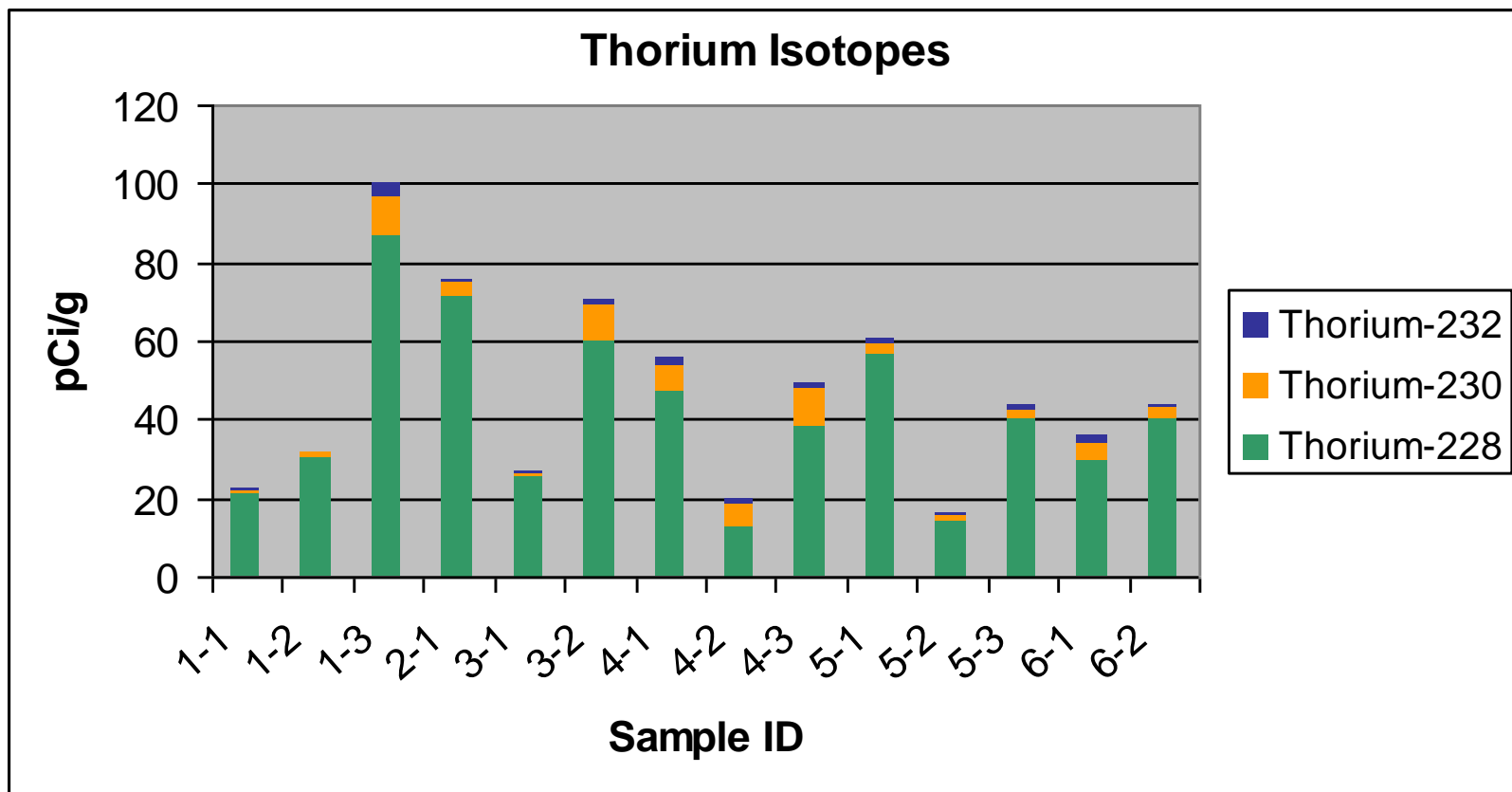
Radium in Distribution System Solids



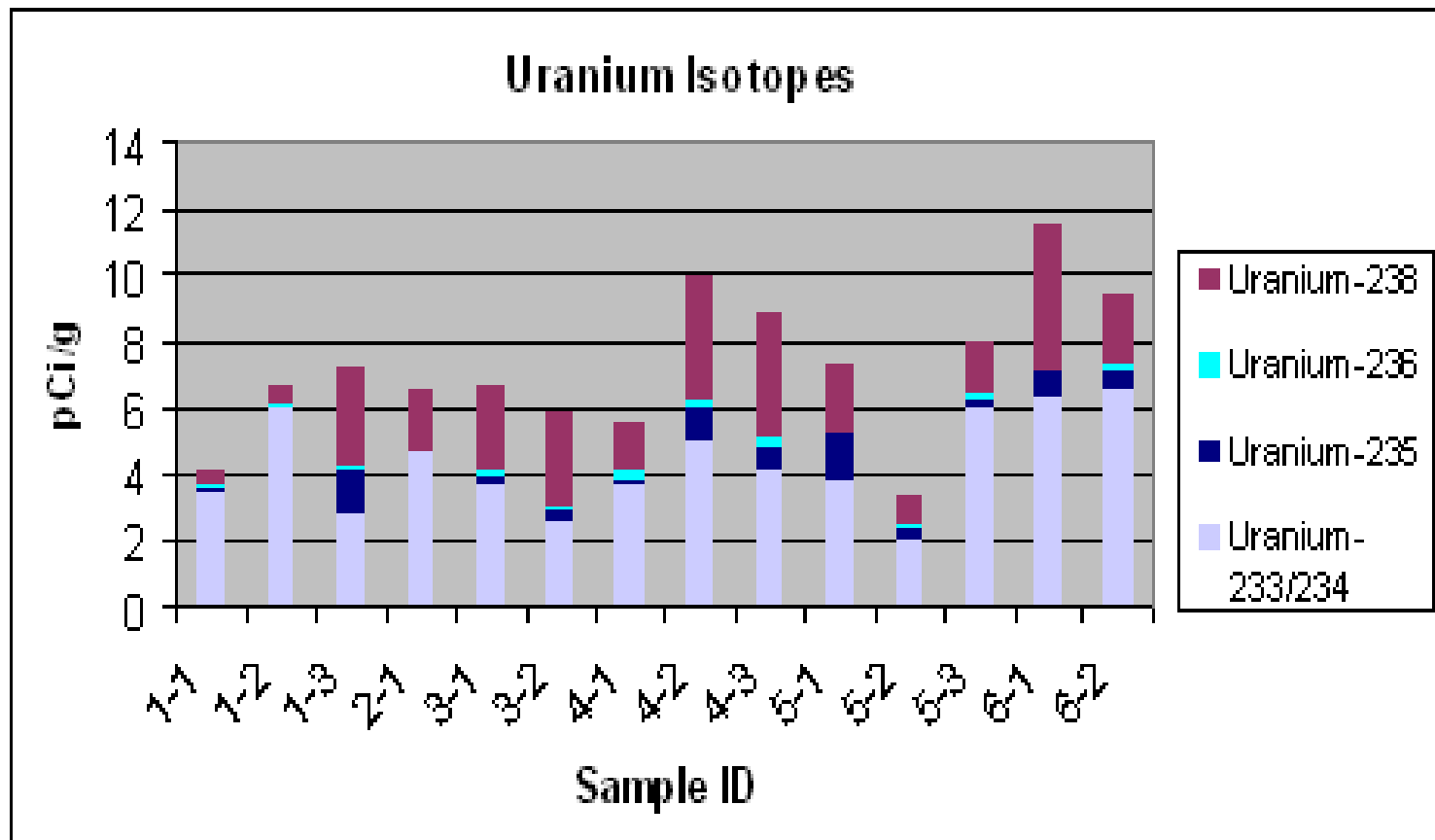
Gross Alpha and Beta Radioactivity in Distribution System Solids



Thorium in Distribution System Solids



Uranium in Distribution System Solids



Conclusions

- Arsenic and some radionuclides can concentrate in DW DS if in water at levels even below respective MCL
- The factors that determine how much accumulates are complex
- Also complex are the factors that impact release back into the distribution system. Disturbances to DS may release arsenic as well as factors such as:
 - *Particle destabilization*
 - *Desorption*
 - *Competitive desorption*
 - *Redox chemistry changes*
 - *Microorganisms*
- Health effects??
- Need for future investigation



Thank-you

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